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Key indicators

Single-crystal X-ray study T = 120 KMean $\sigma(C-C) = 0.004 \text{ Å}$ R factor = 0.047 wR factor = 0.112 Data-to-parameter ratio = 17.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

bu,^a **Maruthai** In the title compound. C

In the title compound, $C_{17}H_{14}O_3S$, there are two independent molecules and the intramolecular dihedral angles between the mean planes of the 4-tolyl and the 2-naphthyl rings are 55.11 (7)° and 65.10 (7)°. $C-H\cdots O$ interactions generate rings of graph-set motif S(5), S(6), $R_1^2(4)$, $R_2^1(6)$, $R_2^1(8)$ and $R_2^1(9)$. The supramolecular aggregation is completed by the presence of several $C-H\cdots \pi$ interactions.

2-Naphthyl 4-toluenesulfonate: supra-

and C—H $\cdots\pi$ interactions

molecular aggregation through C—H···O

enerate 8) and

Comment

p-Toluenesulfonates are used in monitoring the merging of lipids (Yachi *et al.*, 1989), studying membrane fusion during acrosome reactions (Spungin *et al.*, 1992), development of immuno-affinity chromatography for the purification of human coagulation factor (Tharakan *et al.*, 1992), chemical studies on viruses (Alford *et al.*, 1991), development of technology for linking photosensitizers to model monoclonal antibodies (Jiang *et al.*, 1990) and chemical modification of sigma sub-units of the *E. coli* RNA polymerase (Narayanan & Krakow, 1983). An X-ray study of the title compound, (I), was undertaken in order to determine its crystal and molecular structure because of the biological importance of its analogs.



The dihedral angle between the mean planes of the 4-tolyl and 2-naphthyl rings is $65.10 (7)^{\circ}$ for the molecule involving S1, and $55.11 (7)^{\circ}$ for the molecule involving S2 (Fig. 1 and Table 1). This shows their non-coplanar orientation, similar to that found between the 4-tolyl and 2-chlorophenyl rings in 2-chlorophenyl 4-toluenesulfonate (Vembu, Nallu, Garrison & Youngs, 2003*b*) and the 4-tolyl and quinoline rings in 8-tosyloxyquinoline (Vembu, Nallu, Garrison & Youngs, 2003*c*) and in contrast to the near coplanar orientation of the 4-tolyl and 2,4-dinitrophenyl rings in 2,4-dinitrophenyl 4-toluenesulfonate (Vembu, Nallu, Garrison & Youngs, 2003*a*) and 4-tolyl and 4-methoxyphenyl rings in 4-methoxyphenyl 4-toluenesulfonate (Vembu, Nallu, Garrison & Youngs, 2003*a*).

The crystal structure of (I) is stabilized by weak $C-H\cdots O$ interactions. The range of $H\cdots O$ distances (Table 2) agrees with that found for weak $C-H\cdots O$ bonds (Desiraju & Steiner, 1999). The C4-H4 $\cdots O2$, C6-H6 $\cdots O1$, C15-H15 $\cdots O2$ (Fig. 2), C21-H21 $\cdots O4$, C23-H23 $\cdots O5$ and

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Figure 1

The structure of the two independent molecules of (I), showing 50% probability displacement ellipsoids.





C26-H26···O5 intramolecular interactions generate rings of graph-set motif (Etter, 1990; Bernstein et al., 1995) S(5), S(5), S(6), S(5), S(5) and S(6), respectively. The C3-H3···O1^{vii} and C3-H3...O3^{vii} interactions constitute a pair of bifurcated donor bonds, generating a ring of graph set $R_1^2(4)$ (symmetry codes are in Table 2). The H3...O1^{vii} and H3···O3^{vii} distances differ by only 0.04 Å. The resulting configuration is best regarded as a three-centered symmetrical hydrogen-bonded chelate motif (Desiraju, 1989) and is also observed in similar structures (Vembu, Nallu, Garrison & Youngs, 2003b,c,d,e,f,g; Vembu, Nallu, Garrison, Hindi & Youngs, 2003). The C3-H3···O3^{vii} and C1-H1A···O3^{vii} interactions constitute a pair of bifurcated acceptor bonds, generating a ring of graph set $R_2^1(6)$. The C3-H3···O1^{vii} and $C1-H1A\cdots O3^{vii}$ interactions together generate a ring of graph set $R_2^1(8)$. The $R_1^2(4)$ chelate motif and the $R_2^1(6)$ ring motif are present within the $R_2^1(8)$ ring motif. The C21-H21...O5^{iv} and C32-H32...O5^{iv} interactions constitute a pair of bifurcated acceptor bonds, generating a ring of graph set $R_{2}^{1}(9)$. The C24-H24···O4ⁱ and C24-H24···O6ⁱ (Fig. 3) interactions constitute a pair of bifurcated donor bonds, forming another symmetrical three-centered hydrogenbonded chelate motif of graph set $R_1^2(4)$. The H24...O4ⁱ and



Figure 3

View of the crystal structure along the c axis, showing the intermolecular hydrogen bonds.

H24...O6^{*i*} distances differ by 0.07 Å. There are several other weak $C-H \cdots O$ bonds (Table 2, Fig. 3), which contribute to the aggregation of the title compound in the crystal structure. The supramolecular aggregation is completed by the presence of ten C-H··· π interactions (Table 2), where Cg1-Cg6 are the centroids of the rings C2-C7, C8-C9/C14-C17, C10-C13/ C16-C17, C19-C24, C25-C26/C31-C34 and C27-C30/C33-C34, respectively. The details of these $C-H \cdots \pi$ interactions were calculated using PLATON (Spek, 1998).

Experimental

4-Toluenesulfonyl chloride (4.7 mmol) dissolved in acetone (4 ml) was added dropwise to 2-naphthol (5.5 mmol) in aqueous NaOH (2.5 ml, 10%) with constant shaking. The precipitated compound, (I), (3.3 mmol, yield 70%) was filtered off and recrystallized from ethyl acetate.

Crystal data

C ₁₇ H ₁₄ O ₃ S	$D_x = 1.405 \text{ Mg m}^{-3}$
$M_r = 298.34$	Mo $K\alpha$ radiation
Monoclinic, Pc	Cell parameters from 942
a = 7.9315 (6) Å	reflections
b = 29.130(2) Å	$\theta = 2.9-27.4^{\circ}$
c = 6.1111(5) Å	$\mu = 0.24 \text{ mm}^{-1}$
$\beta = 92.383 \ (3)^{\circ}$	T = 120 (2) K
$V = 1410.73 (19) \text{ Å}^3$	Block, colorless
Z = 4	$0.51 \times 0.45 \times 0.26 \text{ mm}$

Data collection

Bruker SMART 6K CCD area- detector diffractometer ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1998) $T_{min} = 0.699, T_{max} = 0.930$ 20090 measured reflections	6450 independent ref 5907 reflections with $R_{int} = 0.045$ $\theta_{max} = 27.5^{\circ}$ $h = -10 \rightarrow 10$ $k = -37 \rightarrow 37$ $l = -7 \rightarrow 7$
Refinement Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.047$ $wR(F^2) = 0.112$ S = 1.04	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.05) + (0.7718P]$ where $P = (F_{o}^{2} + 2)$ $(\Delta/\sigma)_{max} = 0.001$
6450 reflections	$\Delta \rho_{\rm max} = 0.34 \ {\rm e} \ {\rm \AA}^{-3}$

645 379 parameters H-atom parameters constrained

lections $I > 2\sigma(I)$

$541P)^2$ $(F_c^2)/3$ $\Delta \rho_{\rm min} = -0.32 \ {\rm e} \ {\rm \AA}^{-3}$ Absolute structure: Flack (1983), 3197 Friedel pairs Flack parameter = 0.03 (7)

Table 1	
Selected geometric parameters (Å, °).	

1.421 (2)	S1-C5	1.755 (3)
1.426 (2)	O3-C8	1.421 (3)
1.608 (2)	C1-C2	1.504 (4)
120.88 (16)	O4-S2-O5	120.52 (14)
102.82 (13)	O4-S2-O6	102.75 (12)
108.90 (12)	O5-S2-O6	108.85 (11)
109.55 (14)	O4-S2-C22	110.52 (13)
109.36 (15)	O5-S2-C22	109.39 (12)
103.87 (12)	O6-S2-C22	103.24 (11)
118.41 (17)	C25-O6-S2	118.53 (15)
826(2)	$C^{22} = S^2 = O6 = C^{25}$	65 6 (2)
	1.421 (2) 1.426 (2) 1.608 (2) 120.88 (16) 102.82 (13) 108.90 (12) 109.55 (14) 109.36 (15) 103.87 (12) 118.41 (17) 82.6 (2)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Table 2

Hydrogen-bonding geometry (Å, °).

$D - \mathbf{H} \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C24-H24\cdots O6^{i}$	0.95	2.77	3.663 (3)	157
$C24-H24\cdots O4^{i}$	0.95	2.70	3.458 (4)	138
C18-H18A····O4 ⁱⁱ	0.98	2.86	3.495 (4)	124
C18−H18C···O4 ⁱⁱⁱ	0.98	2.38	3.298 (4)	156
$C32-H32\cdots O5^{iv}$	0.95	2.63	3.549 (4)	163
$C21 - H21 \cdots O5^{iv}$	0.95	2.90	3.273 (3)	105
$C6-H6\cdots O2^{iv}$	0.95	2.88	3.138 (3)	97
$C6-H6\cdots O1^{v}$	0.95	2.82	3.555 (4)	135
$C1-H1B\cdots O1^{vi}$	0.98	2.66	3.386 (4)	131
$C1-H1A\cdots O3^{vii}$	0.98	2.85	3.753 (4)	153
C3-H3···O3 ^{vii}	0.95	2.69	3.554 (3)	152
$C3-H3\cdots O1^{vii}$	0.95	2.73	3.521 (4)	142
C26-H26···O5	0.95	2.80	3.111 (3)	100
C23-H23···O5	0.95	2.51	2.900 (3)	104
C21-H21···O4	0.95	2.77	3.054 (4)	98
C15−H15···O2	0.95	2.78	3.012 (4)	95
C6-H6···O1	0.95	2.81	3.061 (4)	96
$C4-H4\cdots O2$	0.95	2.51	2.895 (4)	104
$C7-H7\cdots Cg1^{viii}$	0.95	2.84	3.578	133
$C10-H10\cdots Cg6$	0.95	2.74	3.383	121
$C11-H11\cdots Cg5$	0.95	3.14	3.835	121
$C14-H14\cdots Cg5^{ix}$	0.95	2.74	3.430	128
$C15-H15\cdots Cg6^{ix}$	0.95	3.05	3.744	123
$C20-H20\cdots Cg4^{x}$	0.95	2.82	3.574	138
$C26-H26\cdots Cg3^{vii}$	0.95	3.15	3.806	128
$C27 - H27 \cdots Cg2^{vii}$	0.95	2.63	3.415	132
$C30-H30\cdots Cg2^{iv}$	0.95	3.23	3.956	138
$C31 - H31 \cdots Cg3^{iv}$	0.95	2.84	3.582	132

Symmetry codes: (i) x - 1, y, z; (ii) x - 1, y, z - 1; (iii) $x - 1, 1 - y, z - \frac{1}{2}$; (iv) x, y, z - 1; (v) $x, 2 - y, z - \frac{1}{2}$; (vi) $1 + x, 2 - y, z - \frac{1}{2}$; (vii) 1 + x, y, z; (viii) $x, -y, z - \frac{1}{2}$; (ix) x - 1, y, 1 + z; (x) $x, 1 - y, z - \frac{1}{2}$.

H atoms were located from a difference Fourier map and refined isotropically. The C-H bond lengths are in the range, with isotropic

displacement parameters 1.2 to 1.5 times $U_{\rm eq}$ of the parent atom 0.95–0.98 Å.

Data collection: *SMART-NT* (Bruker, 1998); cell refinement: *SMART-NT*; data reduction: *SAINT-NT* (Bruker, 1998); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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References

- Alford, R. L., Honda, S., Lawrence, C. B. & Belmont, J. W. (1991). Virology, 183, 611–619.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- Bruker (1998). SMART-NT and SAINT-NT. Versions 5.0. Bruker AXS Inc., Madison, Wisconsin, USA.
- Desiraju, G. R. (1989). Crystal Engineering: The Design of Organic Solids. Amsterdam: Elsevier.
- Desiraju, G. R. & Steiner, T. (1999). The Weak Hydrogen Bond in Structural Chemistry and Biology. New York: Oxford University Press.
- Etter, M. C. (1990). Acc. Chem. Res. 23, 120-126.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Jiang, F. N., Jiang, S., Liu, D., Richter, A. & Levy, J. G. (1990). J. Immunol. Methods, 134, 139–149.
- Narayanan, C. S. & Krakow, J. S. (1983). Nucleic Acids Res. 11, 2701-2716.
- Sheldrick, G. M. (1998). SADABS and SHELXTL. University of Göttingen, Germany.
- Spek, A. L. (1998). PLATON. Utrecht University, The Netherlands.
- Spungin, B., Levinshal, T., Rubenstein, S. & Breitbart, H. (1992). *FEBS Lett.* **311**, 155–160.
- Tharakan, J., Highsmith, F., Clark, D. & Drohsn, W. (1992). J. Chromatogr. 595, 103–111.
- Vembu, N., Nallu, M., Garrison, J., Hindi, K. & Youngs, W. J. (2003), Acta Cryst. E59, 0830–0832.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003a), Acta Cryst. E59, 0378–0380.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003b), Acta Cryst. E59, 0503–0505.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003c), Acta Cryst. E59, 0776–0779.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003d), Acta Cryst. E59, 0936–0938.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003e), Acta Cryst. E59, 0939–0941.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003f), Acta Cryst. E59, 01009–01011.
- Vembu, N., Nallu, M., Garrison, J. & Youngs, W. J. (2003g), Acta Cryst. E59, 01019–01021.
- Yachi, K., Sugiyama, Y., Sawada, Y., Iga, T., Ikeda, Y., Toda, G. & Hanano, M. (1989). Biochim. Biophys. Acta, 978, 1–7.